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BOOK OF ABSTRACTS

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CORRELATION BETWEEN METHODS OF ACTIVATION OF MONTMORILONITE K10 AND FORMALDEHYDE CONTENT IN UREA-FORMALDEHYDE COMPOSITES

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ABSTRACT:

In this paper, the methods of activation of montmorillonite (MMT) K10 with the content of formaldehyde (FA) in urea-formaldehyde (UF) composites are compared. MMT K10 samples were activated in two ways: thermally(physical activation) and with sulfuric acid (chemical activation). In situ synthesis of UF composites with unmodified and activated K10 under the same conditions was performed. A total of three UF composites were synthesized: reference UF composite containing unmodified MMT UF/K10, UF composite containing thermally activated MMT (UF/\subseteq K10), and UF composite containing activated MMT with sulfuric acid (UF/ SKK10). Characterization of inactivated and differently activated MMTs as well as synthesized UF/MMT composites was performed using scanning electron microscopy (SEM), X-ray analysis (XRD), thermogravimetric analysis-TG, and infrared spectroscopy (FTIR), and the degree of success of MMT activation was monitored by determining the specific surface area (SP) of MMT using the Sears method. Based on the obtained values for SP, it can be concluded that successful activation of K10 was performed in both ways, because the value increased in both samples compared to unmodified K10 from 74.2 m²/g to 317.4 m^2/g for $\Box TK10$ and 183 m^2/g for SKK10. The content of free FA was determined by the bisulfite method. The obtained results show that there is a correlation between the way of activating MMT K10 and the FA content in the tested UF/MMT composites. The lowest FA content was determined for UF/SKK10 composites and is 0.06% compared to the UF/K10 sample containing 0.5% FA and UF/ \Box TK10 has 0.9% free FA.

Keywords: Activation, Specific surface area, Montmorillonite, free formaldehyde, urea-formaldehyde composite

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